## MDH Protocol:

1.0 ml Real Time Assay (OAA -> Malate)

This is the basic protocol for running a 1.0 ml , continuous enzyme assay (AKA real time) measuring the reduction of $\mathrm{NAD}^{+}$at 340 nm . One unit oxidizes one $\mu$ mole of NADH per minute at $25^{\circ} \mathrm{C}$ and pH 7.4 under the specified conditions. See the introduction to enzyme assay handout for a more complete description of the background of and the performing an enzymatic assay.
Oxaloacetate + NADH + H+ <-> Malate + NAD

Stock Solutions: Solutions must be made fresh each day. Prepare $25 \%$ more than you calculate you will need. If only measuring one or two samples, one ml of OAA and NADH should suffice. For more complicated assays such as specific activity or other kinetic assays, predict your volume of OAA and NADH needed and prepare 25\% more of each solution.

- For all powders stored at $-\mathbf{2 0 ^ { \circ }} \mathbf{C}$ allow bottle to equilibrate at room temperature $\mathbf{1 0} \mathbf{m i n}$ before opening as not to let water condense on the material.
- Use commercial MDH or MDH frozen in glycerol to test your enzyme assay cocktail.

50 mM Na Phosphate buffer (MDH Assay Buffer) pH 8.0 Can be made and stored at RT
Some use 10-50 mM Tris-Cl pH7.4-8.0
20 mM OAA ( $\mathrm{MW}=131.1 \mathrm{amu}$ ) in MDH Assay Buffer. Must be made fresh each day ( $2.62 \mathrm{mg} / \mathrm{ml}$ of buffer)

- Mass approximately 0.02-0.05 g OAA. OAA vol $(\mathrm{ml})=((\mathrm{g}$ massed $/ 131) / 0.02) \times 1000$

10 mM NADH (MW = 704 amu ) in MDH Assay Buffer. Must be made fresh each day ( $7.01 \mathrm{mg} / \mathrm{ml}$ of buffer)

- Mass approximately 0.02-0.05 g NADH. NADH vol (ml) $=((\mathrm{g}$ massed $/ 704) / 0.01) \times 1000$
- With a spectrophotometer zeroed at 340 nm , mix $10 \mu \mathrm{l}$ of NADH and $990 \mu \mathrm{l}$ of MDH Assay Buffer. The OD should be close to 0.6 AU. IF higher, dilute appropriately. If to low, add a small amount of NADH needed to achieve the correct OD.

Store all of the above compounds except Pi buffer in ICE to prevent breakdown of solutions.
DO NOT PRE-MIX NADH and OAA until time to assay. OAA will degrade even at 4oC after ~2 hours. Make Fresh.

## Procedure/Important notes:

- Allow the temperature to equilibrate prior to addition of enzyme.
- Zero spectrophotometer at 340 nm with MDH Assay Buffer.
- Use semi-micro 1.5 ml , plastic, polystyrene cuvettes
- After all components are added, the final volume should be 1.0 ml
- The initial absorbance should 0.6 AU (substituting buffer for enzyme). If not, adjust NADH vol.
- Remember that a flat line means either there is no activity or all of the substrate is gone. In the first case, the absorbance will remain the same as before enzyme addition; in the latter, the absorbance should be significantly decreased (much more than half of the initial absorbance). Always be aware of the magnitude of the OD@340 nm

|  | Volume of Stock | Final Concentration |
| :--- | :--- | :--- |
| MDH Assay Buffer | $970 \mu \mathrm{l}$ or QS to final vol of 1.0 ml |  |
| OAA ( 20 mM ) | $10 \mu \mathrm{l}$ | $200 \mu \mathrm{M}$ |
| NADH $(10 \mathrm{mM})$ | $10 \mu \mathrm{l}$ | $100 \mu \mathrm{M}$ |
| Enzyme ${ }^{* * *}$ | $10 \mathrm{l}^{* * *}$ initiate with enzyme in spec! | $0.005-0.01 \mu \mathrm{~g} / \mathrm{ml}^{*}$ |

* for active wt enzyme $0.5 \mu \mathrm{~g} / \mathrm{ml}$ to $0.1 \mathrm{mg} / \mathrm{ml}$ starting protein conc...

Initiate Assay by addition of enzyme. Be careful to mix well with stir rods or with disposable pipette. The assay should be linear for at least $30-60 \mathrm{sec}$. Ignore the first few seconds as those times are due to mixing. If the $\Delta O D / \mathrm{min}$ is not linear, dilute enzyme with MDH Assay Buffer as necessary and re-assay.

- A non-linear - too fast of a rate indicates that rate is not in first order kinetics and not following requirements to use MM equation!

Determining Km/Vmax: For NADH range from $10-200 \mu \mathrm{M}$ (the high end absorbance may be too high depending on instrument) and OAA range from $10-2000 \mu \mathrm{M}$.

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## Calculating enzyme units

- Enzyme Unit is defined as the rate at which an enzyme catalyzes the conversion of $1 \mu$ mole of substrate to product/minute
- Enzyme Activity is the enzyme units per ml . $1 \mu \mathrm{~mole}$ of substrate to product/minute $/ \mathrm{ml}$
- Specific Activity (SA: a slightly different value) takes in account the activity per mass of protein. Thus SA=1 $\mu$ mole of substrate to product $/ \mathrm{min} / \mu \mathrm{g}$ of enzyme. This calculation is performed when the concentration of the pure protein or in a purification to indicate changes in SA. To calculate SA, first start with the Enzyme Units then divide by $\mu \mathrm{g}$ of protein added to the assay.

To calculate the units in any spectrophotometric-based assay, Beer's Law is used.

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\mathrm{A}=\varepsilon \mathrm{Cl}
$$

Where $\quad A=$ absorbance $\left(\mathrm{M}^{-1} \mathrm{~cm}^{-1}\right)$,
I = pathlength of the cell ( 1 cm for a 1 ml cuvette and 0.585 for $225 \mu \mathrm{l}$ in a 96 plate well),
$C=$ conc of the absorbing species $(M)$
$\varepsilon=$ the molar ext coefficient $\left(\mathrm{M}^{-1} \mathrm{~cm}^{-1}\right)$.
When assaying enzyme activity we measure the slope of the enzyme reaction. The slope of your Absorbance vs time enzyme assay graph is the change in absorbance over time (seconds or minutes). Using minutes, this means we are measuring $\Delta \mathrm{A} / \mathrm{min}$ (change in absorbance per time).

Thus: $\quad \Delta \mathrm{A}=\varepsilon \mathrm{I}(\Delta \mathrm{C})$ - as the concentration of chromophore changes so will the absorbance.
Beer's law can be arranged to determine chance in concentration per time (minute) as a function of absorbance:

$$
\begin{array}{ll}
\Delta \mathrm{A} / \min =\varepsilon \mathrm{I}(\Delta \mathrm{C} / \min ) & \text { adds in the time factor } \\
\Delta \mathrm{C} / \min =(\Delta \mathrm{A} / \min ) /(\varepsilon \times \mathrm{I}) & \text { rearrange factors }
\end{array}
$$

## Calculating Enzyme Activity using NADH oxidation in a 1 ml assay:

NADH has an extinction coefficient of $6.22 \mathrm{mM}^{-1} \mathrm{~cm}^{-1}$

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\(\Delta \mathrm{C} / \min =(\Delta \mathrm{A} / \min ) /(\varepsilon \times \mathrm{I})\)
\(\Delta \mathrm{C} / \mathrm{min}=(\Delta \mathrm{A} / \mathrm{min}) /\left(6.22 \mathrm{mM}^{-1} \mathrm{~cm}^{-1} \times 1.0 \mathrm{~cm}\right)\)
\(\Delta \mathrm{C} / \mathrm{min}=\Delta \mathrm{A} / \mathrm{min} \times 0.161 \mathrm{mM}\)
\(\Delta \mathrm{C} / \mathrm{min}=\) change in absorbance \(/ \mathrm{min}\) (slope of Abs vs Time) \(\times(0.161 \mathrm{mM})=\mathrm{mM} / \mathrm{min}\)
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This is the change in mM of NADH/minute, not the final unit of enzyme activity but close... We want moles not Molarity!

## Next convert $\mathrm{mM} / \mathrm{min}$ to $\mu \mathrm{mole} / \mathrm{min}$, then mmole/min to $\mu \mathrm{mole} / \mathrm{min}$ of NADH converted/min

Using the molarity equation: $M=$ mole/liter ; $\mathrm{mM}=\mu \mathrm{mol} / \mathrm{ml}$
Substituting $\mu \mathrm{mol} / \mathrm{ml}$ for mM we get $\Delta \mathrm{C} / \mathrm{min}=\Delta \mathrm{A} / \mathrm{min} \times 0.161(\mu \mathrm{~mol} / \mathrm{ml}) \mathrm{X}$ final assay volume $(\mathrm{ml})=\mu \mathrm{mol} / \mathrm{min}$ Enzyme Units
For a 1.0 ml assay $\Delta \mathrm{C} / \mathrm{min}=\Delta \mathrm{A} / \mathrm{min} \times(0.161 \mu \mathrm{~mol} / \mathrm{ml}) \times 1.0 \mathrm{ml}=\Delta \mathrm{A} / \mathrm{min} \times 0.161 \mu \mathrm{~mol}=\mu \mathrm{mol}$ of substrate per $\mathrm{min}=$ is enzyme unit.
To get Enzyme Activity (units/ml or $\mu \mathrm{mol} / \mathrm{min} / \mathrm{ml}$ of enzyme), divide Enzyme Unit by the volume of enzyme in ml added to the cuvette ( 0.01 ml for this MDH assay).
$\Delta \mathrm{C} / \mathrm{min}=(\Delta \mathrm{A} / \mathrm{min} \times(0.161 \mu \mathrm{~mol} / \mathrm{ml}) * 1 \mathrm{ml}) / .01 \mathrm{ml}=\Delta \mathrm{A} / \mathrm{min} \times 16.1$ ( $\mu \mathrm{mol}$ of substrate converted per min per ml)

## Conversion from absorbance/time to Enzyme Activity (in min)

1) IF units are in $\Delta \mathrm{Abs} / \mathrm{sec}$, convert to $\Delta \mathrm{Abs} / \mathrm{min}$ by dividing by 60 .
2) $\Delta \mathrm{Abs} / \mathrm{min}$ (the slope of abs $v$ time) $\times 16.1=\mu \mathrm{mol}$ NADH converted $\mathrm{min}^{-1} \mathrm{ml}^{-1}$

## Conversion of Enzyme Activity to Specific Activity

1) Divide the Enzyme Activity by the $\mu \mathrm{g}$ of protein added to cocktail. This is total protein added to the cuvette, not the concentration or the volume.
